Rocket Words		
	thermometer	
2-	melting point	
*** *** ***	freezing point	
	boiling point	
	solid	
S.	liquid	
	gas	
1 1 1	evaporation	
11	particles	
	condensation	
	water vapour	
	substance	

## States of matter

Everything in our universe is made of matter. There are 3 states of matter:



Solid

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Liquid

Solid particles have strong bonds so solids have a fixed shape. Liquid particles have weaker bonds and more energy so liquids can change shape. Gas particles have really weak bonds so gases can spread out and move freely.

Gas

## Condensation



When water vapour (gas) touches a cold surface, the particles lose energy and the bonds become stronger, turning the gas into a liquid.

Careers connected to States of Matter: Chemical Engineer, Pharmacologist, Pharmaceutical pharmacist, Chemist.



Gas

Water Vapour



Heating liquid water increases the particle's energy and the bonds become weaker, turning it into a gas. The hotter the temperature, the faster the rate of evaporation.

**Changes of state** 

Liquid

Water

Different substances have different melting, freezing

In water, the melting and freezing point is 0°C and

Substances can be **heated** or **cooled** to change

States of matter can change.

Solid

Ice

the boiling point is 100 °C.

and boiling points.

from one state to another.

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Solid Liquid Gas	Tick the correct statements.		
Gas particles have lots of energy.		There are strong particle bonds in liquids.	
Solids are a fixed shape.		Solid particles do not have much energy.	
Liquids cannot change shape.		lce is a liquid.	
Gases cannot be squashed.		Helium is a solid.	

Draw lines to match the labels to the thermometers:



Add the following labels to the diagram:



You have been asked to design an experiment to see whether temperature affects the rate of evaporation. What is the variable you will change?

